

Home Search Collections Journals About Contact us My IOPscience

Transport properties of ${\rm FeCl}_3{\rm -doped}$ polypyrroles at different dopant concentrations

This article has been downloaded from IOPscience. Please scroll down to see the full text article. 1994 J. Phys.: Condens. Matter 6 10499 (http://iopscience.iop.org/0953-8984/6/48/011)

View the table of contents for this issue, or go to the journal homepage for more

Download details: IP Address: 171.66.16.179 The article was downloaded on 13/05/2010 at 11:26

Please note that terms and conditions apply.

Transport properties of FeCl₃-doped polypyrroles at different dopant concentrations

A Bhattacharya[†], A De[†]§, S N Bhattacharya[†] and S Das[‡]

 † Nuclear Chemistry Division, Saha Institute of Nuclear Physics, Sector 1, Block-AF, Bidhannagar, Calcutta-700 064, India
 ‡ Solid State and Molecular Physics Division, Saha Institute of Nuclear Physics, Sector 1, Block-AF, Bidhannagar, Calcutta-700 064, India

Received 2 June 1994, in final form 3 August 1994

Abstract. The electrical conductivities of FeCl₃-doped polypyrole samples with different dopant concentrations have been measured in the temperature range 25-300 K. Thermoelectric power measurements have been performed on two polypyrole samples at low doping levels between 50 and 300 K. The variation in DC conductivity can be explained by Mott's variable-range hopping (VRH) model. Fitting with Mott's VRH theory, the conductivity data provide the values of the density $N(\epsilon_{\rm F})$ of states at the Fermi level and the hopping distance R. The low value and the sublinear character of the thermoelectric power is consistent with the VRH theory.

1. Introduction

Recently the synthesis and characterization of conducting polymers [1] have received much attention from scientists because of the possibility of their wide application in solid state devices [2]. The conductivity of this intrinsically insulating material can be varied over a wide range by doping and is influenced by the dopant, the doping level, the synthesis processes and the temperature range of measurement [3-11]. Polypyrrole is a conjugated polymer based on an aromatic ring with a band of 2.5 eV. It has a conduction band of 3.8 eV created by the delocalized electrons of the unpaired $2p_z$ orbitals of the carbon atoms in the chain and has non-degenerate ground states [12]. Doped polypyrrole has similarity with a three-dimensional disordered system in respect of the structure and morphology. Theoretical [13-15] and experimental [16-19] studies provide evidence for the creation of polaron and bipolaron defects in polypyrrole by doping. The polaron state having $\frac{1}{2}$ spin is characterized by an ESR signal due to the enhancement of the spin concentration in *in-situ* ESR measurements [17] and by three optical absorption peaks [18] at 0.7, 1.4 and 2.1 eV within the band region. Two of the optical peaks appear from the transition between the valence band and the bonding and antibonding levels whereas the transition between the bonding and antibonding polaron levels gives rise to a third peak. The third optical peak is absent in the bipolaron state since the bonding level remains vacant. Moreover, the ESR signal will be absent because the spin of the bipolaron state is zero. Up to the intermediate doping level, polaron formation has recently been established [18] but, as the level of doping increases, both polarons and bipolarons may also be formed. A decisive conclusion about the transport mechanism in polypyrrole has not yet been drawn owing to the persistent structural disorder. It may be due to polarons or bipolarons in the polymer chain, to a

[§] Author to whom correspondence should be addressed.

transition between chains, to hopping between localized states or to fluctuation-induced tunnelling, depending upon the temperature range or the existence of metallic regions. At present the highest room-temperature conductivity ($\sigma_{RT}(max) > 10^5$ S cm⁻¹) is obtained in the case of stretched oriented polyacetylene film doped with FeCl₃ and iodine [20, 21]. Compared with FeCl₃-doped polyacetylene, FeCl₃-doped polypyrrole has not attracted much attention. Pron *et al* [22] have reported that polypyrrole can be prepared chemically using the FeCl₃ method. They assigned the anion formed in the reaction as a high-spin Fe(II) species FeCl₄ from the room-temperature Mössbauer spectroscopic studies. The isomer shift is identical and typical of FeCl₄. The motivation for this study is to investigate the temperature dependences of the conductivity and thermoelectric power (TEP) of polypyrrole doped with FeCl₃ and to draw a correlation between the variation in conduction mechanism with the doping level.

2. Experimental details

2.1. Preparation of polypyrrole samples

Pyrrole (E Merck, Germany) employed as a monomer is distilled, purified and kept in an inert atmosphere prior to use. Polypyrrole samples having different dopant concentrations were prepared using different concentrations of FeCl₃ (SD Fine Chemicals Ltd) solutions namely 0.75 M, 0.5 M, 0.25 M, 0.1 M and 0.04 M. The pyrrole concentration was always maintained at 0.05 M. To the 100 ml FeCl₃ solutions in a beaker pyrrole was added dropwise with constant stirring. Polymerization took place almost instantly at room temperature. After about $1\frac{1}{2}$ h the black precipitate of polypyrrole was separated from the solution. The precipitate was washed thoroughly with water until complete removal of the adhered Fe³⁺ and then dried under vacuum for 25 h.

2.2. Measurement technique of resistivity and thermoelectric power

The pellet of polymer was made by pressing in a Perkin–Elmer die. The pressed pellets were used to measure the electrical conductivity in the temperature range 25–300 K. The temperature-dependent resistivities of the samples were measured with the standard four-probe technique using silver paint contacts. A constant DC (about 1–3 mA) from a Lakeshore cryotonics model 120 constant-current source was passed through the sample and the voltage drop across the sample was measured with a Keithley model 181 nanovoltmeter. The current level was kept low to avoid heating of the electrical contact. The low-temperature measurement was performed using an Oxford model 22 closed-cycle helium refrigerator. The sample temperature was measured with a chromel–alumel thermocouple placed very near the sample. A calibrated Si diode was attached to the sample holder to act as the sensor. The temperature was controlled using a Lakeshore cryotonics model DTC500A temperature controller.

The TEP of polypyrole samples was measured using a differential technique where a temperature gradient was created across the sample and the voltage ΔE developed between the hot and cold ends of the thermocouple formed by the sample and Cu wires was measured. Two heaters were used (one at the top and the other at the bottom of the sample) so that either end of the sample may be heated with respect to the other. The temperature difference ΔT between two ends of the sample was kept in the range ± 0.5 K throughout the measured temperature range (77-300 K). The temperature and temperature difference across the sample were measured with chromel-alumel thermocouples. Pressed pellets of

polypyrrole were mounted by pressure contact between the copper plates using silver paint and the whole system was placed in a glass Dewar. The sample temperature was controlled with a Si diode and a manganin heater. At each measurement the temperature of the sample was fixed using a temperature controller (Lakeshore cryotonic model DRC-93CA). At a particular temperature the values of ΔE and ΔT were recorded with the x-y recorder and the TEP S of the sample was calculated from the slope of the ΔE versus ΔT curve using the following relation:

$$\frac{\Delta E}{\Delta T} = \frac{S_{\rm Cu} - S}{S_{\rm chromel} - S_{\rm alumel}}.$$

3. Results and discussion

Mott [23] and Mott and Davis [24] (see also [25]) proposed a model for the phononactivated variable-range hopping (VRH) between localized states irrespective of the presence and absence of polarons. Electrons interact very strongly with the dopant ions in the polymer crystal through an electrostatic interaction. Thus the polymer lattice itself will become deformed by the presence of the electrons. This results in the formation of polarons in localized states. Our data can be fitted much more reliably with the VRH model according to which the conductivity can be expressed as

$$\sigma = \sigma_0 \exp[-(T_0/T)^{1/n}] \tag{1}$$

where the exponent n gives the dimensionality of the charge transport in the solid. The value of n is 3 for two-dimensional and 4 for three-dimensional transport. T_0 is the characteristic temperature.

In the case of three-dimensional transport, T_0 and σ_0 are expressed as [26, 27]

$$T_0 = \frac{\lambda \alpha^3}{k N(\epsilon_{\rm F})} \tag{2}$$

and

$$\sigma_0 = e^2 R^2 \nu_0 N(\epsilon_{\rm F}) \tag{3}$$

where R is the average hopping distance, v_0 is the phonon frequency, k is the Boltzmann constant, α is the inverse localization length and $N(\epsilon_F)$ is the density of states at the Fermi level. Here λ represents a dimensionless constant (about 18.1) [26, 28–30].

The hopping distance R may be expressed as

$$R = \left(\frac{9}{8\pi\alpha kTN(\epsilon_{\rm F})}\right)^{1/4}.$$
(4)

Considering the VRH model, Epstein *et al* [31] gave an expression relating σ and T in three dimensions as

$$\sigma = K_0 T^{-1/2} \exp[-T_0/T)^{1/4}]$$
(5)

10502 A Bhattacharya et al

where

$$T_0 = 16\alpha^3 / kN(\epsilon_{\rm F}) \tag{6}$$

and

$$K_0 = 0.39[N(\epsilon_{\rm F})/\alpha k]^{1/2} v_0 e^2.$$
⁽⁷⁾

We have studied five samples prepared with different concentrations of FeCl₃, namely x = 0.04 M, 0.1 M, 0.25 M, 0.5 M and 0.75 M, which are referred to hereafter as samples A₁, A₂, A₃, A₄ and A₅, respectively. The temperature-dependent behaviour of the resistivity was studied in the temperature range 25-300 K and the TEPs of samples A₁ and A₂ were measured in the temperature range 50 K < T < 300 K. In the case of our polycrystalline samples we obtain a conductivity much less than that obtained by others for polypyrrole films. The properties of polycrystalline material are different from those of single crystals. This property variation is generally attributed to the grain boundaries, i.e. to the localized defects which separate the grains of different orientations, rather than to the differences in the orientations themselves [32-34]. The comparatively large value of resistivity obtained for our samples may arise from the random orientation of microcrystals and their weak coupling through the grain boundary. The influence of the grain boundaries may also have a marked effect on our TEP measurements. Considering equation (1) for $\sigma(T)$ in Mott's VRH theory the activation energy may be expressed as

$$E = -\frac{\mathrm{d}(\ln \sigma)}{\mathrm{d}(1/kT)} = -k\frac{1}{n}T_0^{1/n}T^{-(1/n-1)}.$$
(8)



Figure 1. Hill plot showing the change in $\ln E$ (E is the activation energy) as a function of $\ln T$.



Figure 2. Variation in $\ln \sigma$ as a function of $T^{-1/4}$ for different samples.

A Hill [35] plot (ln E versus ln T curve) is shown in figure 1 which is a straight line of gradient -(1/n-1). This plot actually presents the effect of the different levels of doping on the conductivity. The value of n may change from n = 4 for various reasons such as variation in dimensionality, spatial inhomogeneities or variation in the density of states with energy. From the gradient of the straight line the value of n is determined. It is found that for all the samples the value of n is 4. So the Hill plot indicates that the three-dimensional VRH mechanism plays an important role in the conduction process in our samples. We analysed our conductivity data in the light of the VRH theory with n = 4. In figure 2 we plotted $\ln \sigma$ versus $T^{-1/4}$ together with the theoretical fit curve. It is observed that the curve is linear for all the samples although the range of temperatures for linearity varies. To avoid unrealistic parameters we used guessed values of T_0 and σ_0 for fitting from the slope and intercept, respectively, of the experimental conductivity curves presented in figure 2. Table 1 represents the different parameters obtained by fitting with Mott's VRH theory. Here we have assumed that the localization length α^{-1} of localized electrons is about four pyrrole rings [15]. From the calculation of the energetics of polaron and bipolaron formation on polypyrrole chains, Bredas et al [15] showed that the formation of a bipolaron is favoured over that of two polarons by 0.45 eV and the bipolaron length extends to about four pyrrole rings. According to Ehring and Roth [37] the VRH theory becomes valid when at low temperatures the condition $(T_0/T)^{1/4} \gg 10$ is fulfilled. Mott and Davis derived the VRH theory on the assumption that the wavefunction becomes localized in a small area of the system. The values of $(T_0/T)^{1/4}$ obtained here are 6-10. The small values obtained in this case indicates delocalization of the wavefunction in polypyrrole as seen by Bredas et al [15]. We also fitted the data with Epstein's model, but the fit is not so good as for

Mott's VRH model. Figure 3 presents the $\ln(T^2\sigma)$ versus $T^{-1/4}$ plot. The slope $T_0^{1/4}$ and hence T_0 obtained by fitting with equation (1) are plotted in figure 4. The plot shows a systematic increase in T_0 with increasing dopant concentration. From equation (2) it is seen that the value of T_0 depends largely on the localization length and the density of states at the Fermi level. The variation in these two parameters in the system with different dopant concentrations may induce change in the value of T_0 . The value of $N(\epsilon_F)$ found here is in agreement with the other experimental values obtained by different experimental techniques [38]. It is found that the values of $N(\epsilon_F)$ and R_0 change with the dopant concentration. This may be due to the arbitrary choice of the localization length $\alpha^{-1} = 10$ Å for the samples. From equation (3) the value of the hopping frequency v_0 is calculated and found to be about 10^{13} s⁻¹.



Figure 3. Plot of $\ln(\sigma T^{1/2})$ versus $T^{-1/4}$: ---, theoretical curve fitting with Epstein's formula.

 Table 1. Variable-range hopping parameters for three-dimensional hopping in FeCl3-doped polypyrrole.

Sample	Characteristic temperature T_0 (K)	Density of states $N(\epsilon_{\rm F}) (\alpha^{-1} = 10 \text{ Å})$ $(eV^{-1} \text{ cm}^{-3})$	Hopping distance $R_{3(X)} (\alpha^{-1} = 10 \text{ Å})$ Å
A	169 985	1.236×10^{21}	32.54
A ₂	460 050	4.566×10^{20}	23.47
A ₃	989 474	2.123×10^{20}	28.43
A4	2216717	9.476 × 10 ¹⁹	19.56
As	3 477 484	6.044 × 10 ¹⁹	38.92



Figure 4. Characteristic temperature T_0 versus doping concentration.



Figure 5. Variation in thermopower S with square root $(T^{1/2})$ of temperature.

The TEP measurement is an extremely useful probe for investigating the intrinsic electronic conduction mechanism of the system. The contribution of the hopping mechanism is also suggested by the measurement of the TEP. According to Mott's VRH theory the TEP can be expressed as [24, 39]

$$S(T) = (k^2/2e)(T_0T)^{1/2} \operatorname{d}(\ln N)/\operatorname{d}\epsilon_{\epsilon=\epsilon_s}$$
(9)

i.e. the TEP is approximately proportional to $(T)^{1/2}$. Here T_0 is the parameter appearing in the Mott's VRH law for the conductivity. The TEP at room temperature is 10.20 μ V K⁻¹ for sample A₁ and 9.17 μ V K⁻¹ for sample A₂. The observation of low TEP values is consistent with the VRH theory. The TEP presents a non-linear behaviour with the variation in temperature and in both the cases the TEP decreases with decrease in temperature. In figure 5 we have plotted the TEP against $T^{1/2}$ which gives a linear curve, indicating that the VRH model is applicable here. To bring the two TEP curves into the same figure we have added 0.5 to all the TEP values for sample A₂. From the slope of the curve the value of d(ln N)/d ϵ is determined to be 2.8 eV⁻¹ for sample A₁ and 1.5 eV⁻¹ for sample A₂. In our samples the contribution from hopping becomes predominant and a linear metal-like component is absent. The magnitude of the TEP is largely consistent with the values for semiconductive polymers. It is found that the nature of the TEP does not change with the dopant concentration.

4. Conclusion

We have measured the DC conductivity of FeCl₃-doped polypyrrole in the temperature range 25 K < T < 300 K and TEP between 50 and 300 K. The temperature dependence can be explained well using the VRH developed by Mott and Davis where the polaronic hopping plays the dominant role in the transport process. The parameters obtained from the VRH model are in good agreement with those obtained by other experimental techniques. The contribution of the hopping mechanism is also suggested by the measurement of the TEP. The TEP decreases with decrease in temperature. This observation is in good agreement with that in the case of the interchain conduction by hopping carrier. So Mott's VRH mechanism plays an important role in the conduction mechanism for doped polypyrrole in the insulating regime, and both sets of results presented here appear to be consistent with the model of an amorphous semiconductor.

Acknowledgments

The authors are grateful to Professor B Ghose and other colleagues of the Low Temperature Physics Section of the Saha Institute for rendering help in the resistivity and TEP measurement.

References

- [1] Bredas J L 1986 Handbook of Conducting Polymers vol 2, ed T A Skotheim (New York: Dekker) p 859
- [2] Scoch K F Jr 1986 IEEE Electron. Insul. Magn. 2 20
- [3] Travers J P, Audebert P and Bidan G 1985 Mol. Cryst. Liq. Cryst. 118 B-149
- [4] Bukhs E and Hodges I M 1985 J. Chem. Phys. 83 5976
- [5] Satoh M, Imanishi K, Yashuda Y, Tsushima R, Akoi S and Yoshino K 1988 J. Phys. D: Appl. Phys. 21 1782
- [6] Sato K, Yamaura M, Hagiwara T, Murata K and Tokumoto M 1991 Synth. Met. 44 35
- [7] Singh R, Tandon R P, Panwar V S and Chandra S 1991 J. Appl. Phys. 69 2504
- [8] Singh R, Tandon R P and Chandra S 1993 J. Phys.: Condens. Matter 5 1313
- [9] Roy R, Sen S K, Digar M and Bhattacharya S N 1991 J. Phys.: Condens. Matter 3 7849
- [10] Watanabe A, Tanaka M and Tanaka J 1981 Bull. Chem. Soc. Japan 43 2271
- [11] Castillo-Ortega M M, Inoue M B and Inoue M 1989 Synth. Met. 28 C65
- [12] Street G B 1986 Handbook of Conducting Polymers vol 1, ed T A Skotheim (New York: Dekker) p 265
- [13] Bredas J L, Themans B, Fripiat J G, Andre J M and Chance R R 1984 Phys. Rev. B 29 6761
- [14] Brazovskii S H and Kirova N N 1981 JETP Lett, 33 5
- [15] Bredas J L, Scott J C, Yakushi K and Street G B 1984 Phys. Rev. B 30 1023
- [16] Ford W K, Duke C B and Salocneck W R 1982 J. Chem. Phys. 77 5030
- [17] Genoud F, Gugliemi M, Nechtschein M, Genies E and Salmon M 1985 Phys. Rev. Lett. 55 118
- [18] Zotti G and Schiavon G 1989 Synth. Met. 30 151
- [19] Nechtschien M, Devreax F, Genoud F, Vieil E, Pernaut J M and Genies E 1986 Synth. Met. 15 59
- [20] Naarmann H and Theophilon N 1985 Synth. Met. 22 1
- [21] Tsukamoto J, Takahashi A and Kawasaki K 1990 Jupan. J. Appl. Phys. 29 125
- [22] Pron A, Kucharski Z, Budrowski C, Zagorska M, Krichene S, Suwalski J, Dehe G and Lefrant S 1985 J. Chem. Phys. 83 5923
- [23] Mott N F 1967 Adv. Phys. 16 49; 1969 Phil. Mag. 19 835
- [24] Mott N F and Davis E A 1979 Electronic Processes in Noncrystalline Materials (Oxford: Clarendon) pp 32-4
- [25] Chien J C W 1984 Polyacetylene (Orlando, FL: Academic) p 485
- [26] Paul D K and Mitra S S 1973 Phys. Rev. Lett. 31 1000
- [27] Nair K and Mitra S S 1977 J. Non-Cryst. Solids 24 1
- [28] Ambegaokar V, Halperin B I and Langer J S 1971 Phys. Rev. B 4 2612
- [29] Fritzsche H 1974 Amorphous Semiconductors ed J Tauc (New York: Plenum) p 233

- [30] Broadsky M H and Gambino R J 1972 Proc. 4th Int. Conf. on Amorphous and Liquid Semiconductors (Ann Arbor, MI, 1971) ed M H Cohen and G Lukovsky (Amsterdam: North-Holland) p 742
- [31] Epstein A J, Rommelmann H, Bgelow R, Gibson H W, Hoffman D and Tanner D B 1983 Phys. Rev. Lett. 50 1866
- [32] Amilinckx S and Dekeyser W 1959 Solid State Physics vol 8, ed F Seitz and D Turnbull (New York: Academic) p 325
- [33] Lormand G 1982 J. Physique Coll. 43 C6 283
- [34] Rossiter P L 1987 The Electrical Resistivity of Metals and Alloys (Cambridge Solid State Science Series) ed R W Cahn, E A Davis and I M Ward (Cambridge: Cambridge University Press) p 185
- [35] Hill R M 1976 Phys. Status Solidi 34 601
- [36] Bottger H and Bryskin V V 1985 Hopping Conduction in Solids (Weinheim: VCH) p 155
- [37] Ehring K and Roth S 1986 Phil. Mag. B 53 301
- [38] Pfluger P, Gubler U M and Street G B 1984 Solid State Commun. 49 911
- [39] Kaiser A B 1989 Springer Series in Solid State Sciences vol 91, ed H Kuzmany, M Mehring and S Roth (Berlin: Springer) p 29